## COMPONENTS:

- (1) 1,6-Heptadiyne; C<sub>7</sub>H<sub>8</sub>; [2396-63-6]
- (2) Water; H<sub>2</sub>O; [7732-18-5]

#### ORIGINAL MEASUREMENTS:

McAuliffe, C.

J. Phys. Chem. 1966, 70, 1267-75.

### VARIABLES:

One temperature: 25°C

### PREPARED BY:

- A. Maczynski, Z. Maczynska, and
- A. Szafranski

### EXPERIMENTAL VALUES:

The solubility of 1,6-heptadiyne in water at 25°C was reported to be 1650  $g(1)/10^6$  g(2).

The corresponding mass percent and mole fraction,  $x_1$ , calculated by the compilers are 0.1650 g(1)/100 g sln and 3.23 x  $10^{-4}$ .

# AUXILIARY INFORMATION

## METHOD/APPARATUS/PROCEDURE:

In a 250-mL bottle, 10-20 mL of (1) was vigorously shaken for 1 hr, or magnetically stirred for 1 day, with 200 mL of (2) at 25°C. The bottle was set aside for 2 days to allow droplets of undissolved (1) to separate. Absence of emulsion was checked microscopically. A sample of the hydrocarbon-saturated water was withdrawn with a Hamilton syringe and gas liquid chromatographed in conjunction with a flame-ionization detector.

## SOURCE AND PURITY OF MATERIALS:

- Phillips Petroleum or Columbia Chemical; used as received.
- (2) distilled.

### ESTIMATED ERROR:

temp.  $\pm$  1.5 K soly. 25 g(1)/10<sup>6</sup> g(2) (standard deviation of mean)

## REFERENCES: